



论文

福岛核事故源项评价

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摘要 2011 年日本福岛发生的核泄漏对全球环境和社会产生巨大影响, 核事故的后果评价与环境修复成为长期而复杂的任务, 而源项作为事故分级的重要依据, 是后果评价和环境修复的基础. 本文从福岛核事故泄漏的核素总量与组成、特征核素谱与其他事件的历史排放水平对比等三方面出发, 对福岛核事故的源项进行分析. 分析结果表明, 福岛核事故泄漏放射性总量大约是切尔诺贝利核事故泄漏总量的 10%, 不到全球落下灰总量的千分之一, 但是福岛核事故是至今为止最为严重的海洋放射性污染核事故. 本文还对核事故后处理措施进行评估, 针对福岛核事故的环境影响提出一些建议.

关键词福岛核事故
切尔诺贝利核事故
全球落下灰
放射性
大气
海洋

2011年3月11号日本东部海域发生9级大地震, 随后引发的海啸浪高超过14 m, 破坏福岛第一核电站的冷却系统, 并随即引发堆芯融化、氢气爆炸, 大量放射性物质泄漏进入大气、海洋和陆地环境(Povinec等, 2013c).

日本福岛核事故被国际原子能机构(IAEA)定为7级核事故, 其泄漏放射性物质总量大约为切尔诺贝利核事故的10%~15%(Steinhauser等, 2014), 却是至今为止最严重的海洋放射性污染的核事故(Povinec等, 2013c; Buessler, 2014). 福岛核事故排放的放射性物质影响整个北半球(Thakur等, 2013), 甚至部分南半球地区(Orr等, 2013).

核事故后的源项估计是事故分级的重要依据,

是后果评价的基础. 福岛核事故放射性物质排放存在间断式的事件排放与长时期非点源排放. 福岛核事故之后由于电力与通信的缺失, 核事故初期监测数据的空白, 核事故后期存在一些不可预期的非点源排放, 同时监测站位与频率的限制等, 都给源项估计带来很大挑战.

福岛核事故已过去3年, 国际上已经开展大量的研究, 但较为分散. 本文从源项入手, 大量收集福岛核事故泄漏的放射性物质经大气、海洋途径排放的源项数据; 从历史排放角度出发, 对比福岛核事故、切尔诺贝利核事故、全球落下灰的排放量; 从事故排放的放射性核素组成出发, 寻找日本福岛核事故的指纹特征, 便于今后端元分析, 定量不同源项的核素贡献;

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对同为7级的福岛核事故与切尔诺贝利核事故进行比较;最后针对福岛核事故的源项进行总结及展望。

1 大气排放

核事故后的大气排放是对环境影响最为直接、迅速,影响范围最为广泛的途径。福岛核事故后许多研究者开展大气排放放射性物质总量的评估(Bailly du Bois等, 2014; Cervone和Franzese, 2014; Koo等, 2014; Lin等, 2015b),但是仍然存在部分核素未被统计。全面了解福岛核事故来源的人工放射性核素排放的总量与组成,对于事故分析、辐射评价、环境修复和核素生物地球化学过程研究等具有重要意义(Steinhauser, 2014)。表1给出至今为止最为全面的不同核素通过大气排放的放射性总量估计,由于日本方面发布信息透明度及可信度存疑,监测站位与频率的不足,源项估计时间、空间及方法上的差异等,不同研究者对于同一核素排放总量的估算存在差异,本文一并保留并给出相应出处,以便今后进一步深入的分析与评价。

^{131}I , ^{134}Cs 和 ^{137}Cs 由于其分析方法相对简单,核事故后有大量的环境监测数据,同时三者也是辐射剂量评价中最为主要的核素,因此对于三者的源项估计研究最多。从表1中可以得到 ^{131}I , ^{134}Cs 和 ^{137}Cs 大气排放的源项分别为60~390, 15~20和5~50 PBq。不同大气输运模型结构差异,也将影响源项的估算结果。

2 海洋排放

福岛核事故泄漏大量放射性物质直接进入海洋,是至今为止最为严重的海洋放射性污染的核事故(Lin等, 2015a)。冷却水与堆芯直接接触,除 ^{131}I , ^{134}Cs 和 ^{137}Cs 等低沸点核素外,部分高沸点的 ^{90}Sr 和 $^{239+240}\text{Pu}$ 核素也可能泄露进入海洋环境(Povinec等, 2012; Bu等, 2013a; Casacuberta等, 2013; Zheng等, 2013),核素泄漏总量与成分取决于冷却水酸度与液体排放量。表2总结目前最为全面的不同核素通过海洋途径排放的放射性源项估计。

由于海洋监测复杂性,海洋排放的研究相对于大气排放的研究较少。陆地环境中,通过大气排放的U, Pu, Am和Cm等核素沉降于核电站附近的土壤(Yamamoto等, 2012, 2014a, 2014b; Zheng等, 2012;

Schneider等, 2013; Sakaguchi等, 2014)、植物表面(Shozugawa等, 2012; Schneider等, 2013)和河流沉积物(Evrard等, 2014)和气溶胶(Shinonaga等, 2014)等,来自福岛核事故的核素可以被采样、测量、识别,表1根据陆地环境测量结果,反推福岛核事故来源的超铀元素的泄漏量;海洋环境中也必然存在福岛核事故来源的U, Pu, Am和Cm等核素,然而由于海洋较强的稀释能力,福岛核事故排放的放射性物质进入海洋后迅速被稀释,至今为止海洋环境中还没有来自福岛核事故的超铀核素的直接证据(Bu等, 2013a, 2013b, 2014a, 2014b; Oikawa等, 2014),因此表2并未给出海洋途径排放的超铀元素的泄漏量。

^{131}I , ^{134}Cs 和 ^{137}Cs 作为海洋辐射剂量评价关注的焦点核素,根据表2得到, ^{131}I , ^{134}Cs 和 ^{137}Cs 海洋排放的源项分别为10~61, 2~20和1~30 PBq。 ^{137}Cs 作为一个长半衰期的重要人工放射性核素,其研究结果最为丰富,Buesseler等(2012)的结果是西北太平洋调查海域的 ^{137}Cs 储量,因此2 PBq是非常保守的估计;东京电力公司(TEPCO)基于实测估算2号反应堆中高放射性冷却水中 ^{137}Cs 的泄漏量为4.7 PBq,该结果尚未考虑1~4号反应堆的泄露,地下水排放,人为低放射性废水排放等过程,因此该结果也较为保守;Charette等(2013)根据Buesseler等(2012)的实测结果,结合锕同位素数据估算海洋途径 ^{137}Cs 排放量为11~16 PBq,该结果通过锕获得物理输运过程的参数,由于数据稀缺,存在较大的不确定度;Bailly du Bois等(2012)估算的 ^{137}Cs 通过海洋途径的排放总量为27 PBq,该结果考虑了更多过程的海洋输入途径。根据以上分析及表2结果,作者认为福岛核事故来源的 ^{137}Cs 的海洋途径排放量保守估计为5 PBq。

虽然福岛核事故后,多个放射性物质输运模型曾经进行比较(Masumoto等, 2012),不同模型的不确定性来源存在差异,而且程度也不同,某一个模型的结果很难全面代表源项情况,因此源项评价是一个长期过程,需要利用不同技术、从不同角度开展工作,减少不确定度,增加结果的可信度。

3 源项对比

福岛核事故与切尔诺贝利核事故都被IAEA定为最高的7级核事故,为更好的了解福岛核事故源项排

表 1 大气排放的放射性物质源项估计(10^{15} Bq)

核素	排放总量
^{85}Kr	44.1(Ahlsweide等, 2013) 61(Nitta等, 2014) 83.7(Povinec等, 2013c)
^{90}Sr	0.14(Povinec等, 2013c)
$^{110\text{m}}\text{Ag}$	1.5×10^{-2} ~ 0.15 (Saito等, 2014), 3.0×10^{-2} (Lepage等, 2014)
$^{129\text{m}}\text{Te}$	3.33(Povinec等, 2013c)
^{129}I	8.06×10^{-6} (Hou等, 2013)
^{131}I	65.3(Ten Hoeve和Jacobson, 2012), 105.9(Saunier等, 2013) 120(Nagai等, 2014), 100~400(Achim等, 2012) 151(Katata等, 2012), 153~160(Chino等, 2011) 159(Povinec等, 2013c), 160(Povinec等, 2013c) 190(Mathieu等, 2012), 100~200(Akahane等, 2012) 190~380(Winiarek等, 2012), 320(TEPCO, 2011) 200(IRSIN, 2011; Kobayashi等, 2013) 360~390(ZAMG 2011)
^{132}I	35.8(Saunier等, 2013)
^{132}Te	88(Tagami等, 2013), 88.4(Povinec等, 2013c)
^{133}I	42.2(Povinec等, 2013c)
^{133}Xe	6000(Achim等, 2012), 12000(Bowyer等, 2011; Povinec等, 2013c) 22000(TEPCO, 2011), 12134(Saunier等, 2013) 14000(Stohl等, 2012a), 15300(Stohl等, 2012) 17000(NILU, 2011), 20000(IRSIN, 2011)
^{134}Cs	15(Povinec等, 2013c), 17.5(Povinec等, 2013a) 18(Hamada和Ogino, 2012)
^{135}Cs	6.74×10^{-5} (Zheng等, 2014)
^{137}Cs	5.5~9.7(Miyazawa等, 2012), 8.8(Nagai等, 2014) 11.5(Bailly du Bois等, 2012), 10(Morino等, 2011; Achim等, 2012) 12(Yasunari等, 2011; Winiarek等, 2012) 13(Kawamura等, 2011; Kobayashi等, 2013) 15(Honda等, 2012; Povinec等, 2013c) 15.3(Povinec等, 2013b), 15.5(Saunier等, 2013) 13~15(Chino等, 2011), 17(Ten Hoeve和Jacobson, 2012) 20(Mathieu等, 2012; Korsakissok等, 2013; Bailly du Bois等, 2014) 10~20(Akahane等, 2012) 30(IRSIN, 2011), 36(NILU, 2011), 36.6(Stohl等, 2012b) 50(ZAMG (Central Institute for Meteorology and Geodynamics, 2011)
U	3.9×10^{-9} (Sakaguchi等, 2014)
$^{239+240}\text{Pu}$	1×10^{-6} ~ 2.4×10^{-6} (Zheng等, 2012), 2.3×10^{-6} (Sakaguchi等, 2014) 3.5×10^{-6} (Yamamoto等, 2014b)
^{241}Am	5.0×10^{-7} ~ 1.3×10^{-6} (Yamamoto等, 2014b)
^{242}Cm	2.7×10^{-5} ~ 6.5×10^{-5} (Yamamoto等, 2014b)
$^{243+244}\text{Cm}$	1.1×10^{-6} ~ 2.6×10^{-6} (Yamamoto等, 2014b)

表 2 海洋排放的放射性物质源项估计(10^{15} Bq)

核素	排放总量
^3H	0.1(Povinec等, 2013c)
^{90}Sr	0.09~0.9(Casacuberta等, 2013)
	0.08(Periáñez等, 2013), 1~6(Povinec等, 2012)
$^{99\text{m}}\text{Tc}$	0.02(Nair等, 2014)
^{129}I	2.35×10^{-6} (Hou等, 2013)
	7×10^{-6} (Guilderson等, 2013; Povinec等, 2013a)
^{131}I	11(Kawamura等, 2011), 11.1(Tsumune等, 2013)
	61.6(Hou等, 2013)
^{134}Cs	2.24(Nair等, 2014), 3.5(Tsumune等, 2013)
	11~16(Charette等, 2013)
^{135}Cs	7.01×10^{-5} (Zheng等, 2014)
^{136}Cs	0.29(Nair等, 2014)
^{137}Cs	0.9~3.5(Dietze和Kriest, 2012), 2(Buesseler等, 2012)
	2.2(Povinec等, 2013c), 3.5(Tsumune等, 2012, 2013)
	3.6(Tsumune等, 2013), 4(Kawamura等, 2011)
	5.1~5.5(Estournel等, 2012), 5.5~5.9(Miyazawa等, 2012)
	16.2(Rypina等, 2013), 27(Bailly du Bois等, 2012)
^{140}Ba	0.53(Nair等, 2014)
^{140}La	0.27(Nair等, 2014)

放情况,表3给出全球落下灰、切尔诺贝利核事故、福岛核事故排放的放射性物质对比情况,便于从历史排放水平的角度上,更直观的了解福岛核事故的源项排放情况。

根据表3可知,如果不考虑放射性惰性气体的排放(^{85}Kr 和 ^{133}Xe),福岛核事故来源的 ^{131}I 和 ^{137}Cs 等主

要的放射性核素排放量总量大约为切尔诺贝利核事故的10%,不足全球落下灰的千分之一;如果考虑放射性惰性气体的排放(^{85}Kr 和 ^{133}Xe),福岛核事故泄漏放射性物质总量高于切尔诺贝利核事故,因为福岛核事故共有4个反应堆发生不同程度的堆芯熔化,而切尔诺贝利核事故只有一个反应堆发生堆芯熔化;对于 $^{239+240}\text{Pu}$,福岛核事故泄漏量远小于切尔诺贝利核事故和全球落下灰的排放量。

4 福岛核事故排放的核素谱的特征指纹

不同核事故泄漏的特征核素谱存在差异,取决于核燃料组成、反应堆燃耗、泄漏过程和冷却时间等因素。了解事故后核素活度比值与原子数比值,可以判断污染源、反演历史排放情况、估算其他核素的排放总量、了解堆芯融化情况、评估核燃料燃耗等。对于同一污染源,核素运移过程中存在生物地球化学行为差异、同位素分馏、不同源项的混合、放射衰变等过程,将导致核素比值的改变;对于不同污染源,核燃料组成、反应堆燃耗情况、核燃料泄漏过程及方式、核燃料冷却时间等将影响核素比值。表4对比日本福岛核事故、切尔诺贝利核事故、全球落下灰核素比值,该表有利于今后不同端元的确定,计算不同来源放射性物质含量的贡献。

核素活度比值、原子数比值还可以用来估计其他核素的排放总量,特别是对于一些分析方法较为复杂的核素(^{90}Sr 和 ^{129}I),事故之后由于测量资源的限制(时间、仪器、人力),无法同时大量开展监测,可以

表 3 福岛核事故、切尔诺贝利核事故与全球落下灰放射性物质排放量(10^{15} Bq)

	福岛核事故		切尔诺贝利核事故		全球落下灰	
	大气	海洋	大气	海洋	大气	海洋
^{131}I	160 ^{a)}	11 ^{b)}	1760 ^{j)}			
^{137}Cs	15 ^{a)}	4 ^{b)}	85 ^{j)}	16 ^{m)}	950 ^{j)}	600 ^{m)}
^{90}Sr	0.14 ^{b)}	1 ⁱ⁾	10 ^{j)}		600 ^{j)}	380 ^{m)}
^{133}Xe	$1.2\times 10^4\sim 1.5\times 10^{4\text{c),d)}$		$2.0\times 10^3\sim 6.0\times 10^{3\text{k)}$			
^{85}Kr	44.1 ^{e)}		33 ^{e)}		5000	
$^{239+240}\text{Pu}$	$1.0\times 10^{-6}\sim 2.4\times 10^{-6\text{f)}$		0.1 ^{l)}		10 ^{j)}	
排放总量 ⁿ⁾	520 ^{g)}		5200 ^{l)}		712920 ^{j)}	

a) Povinec等(2013b); b) Povinec等(2013c); c) Bowyer等(2011); d) Stohl等(2012b); e) Ahlswede等(2013); f) Zheng等(2012); g) Steinhauser等(2014); h) Kawamura等(2011); i) Povinec等(2012); j) UNSEAR(2008); k) Ginzburg和Reis(1991); l) Livingston和Povinec(2002); m) IAEA(2005); n) 排放总量中不包含放射性惰性气体(^{85}Kr 和 ^{133}Xe)

表 4 福岛核事故、切尔诺贝利核事故和全球落下灰核素比值

	福岛核事故		切尔诺贝利核事故		全球落下灰	
	活度比值	原子数比值	活度比值	原子数比值	活度比值	原子数比值
$^{134}\text{Cs}/^{137}\text{Cs}$	1 ^{a)-c)}		0.5 ^{a)}			
$^{135}\text{Cs}/^{137}\text{Cs}$ ^{d)-f)}		0.37		0.46		2.7
$^{131}\text{I}/^{137}\text{Cs}$	15 ^{g)}		20			
$^{129}\text{I}/^{131}\text{I}$		22.3 ^{h)}		19 ⁱ⁾		
$^{129}\text{I}/^{137}\text{Cs}$ ^{j)}	4.43×10^{-7}		2.75×10^{-7}			
$^{90}\text{Sr}/^{137}\text{Cs}$	0.01~0.02 ^{b),k)}		0.08 ^{l)}		0.63 ^{b)}	
$^{240}\text{Pu}/^{239}\text{Pu}$ ^{m)}		0.3		0.4		0.18
$^{241}\text{Pu}/^{239}\text{Pu}$		0.103~0.135 ^{m)}				$1.9\times 10^{-3n)$
$^{241}\text{Pu}/^{239+240}\text{Pu}$	107.8 ^{m)}		83 ^{m)}		13~14 ^{o)}	
$^{238}\text{Pu}/^{239+240}\text{Pu}$	2.2 ^{p)}		0.5 ^{o)}		0.033 ^{p)}	

a) Masson等(2011); b) Buessler等(2012); c) Merz等(2013); d) Ohno和Muramatsu(2014); e) Shibahara等(2014); f) Zheng等(2014); g) Hirose(2012); h) Miyake等(2012); i) Kutschera等(1988); j) Tumey等(2013); k) Periañez等(2013); l) Aarkrog(2003); m) Zheng等(2012); n) Kelley等(1999); o) Hirose等(2001); p) Lujanienė等(2012)

通过该核素与其他容易测量的核素(^{137}Cs 和 ^{131}I)比值间接估算事故中的泄漏量;同时核素活度和原子数比值还可以用来判断放射性物质的来源,定量不同来源的源项贡献。

5 福岛核事故与切尔诺贝利核事故比较

福岛核事故与切尔诺贝利核事故的差异从三方面展开讨论:放射性物质总量、放射性物质成分和核素迁移途径。

对于泄漏放射性物质总量,虽然日本福岛核事故有4个反应堆发生氢气爆炸,并伴随着不同程度的放射性物质泄漏,切尔诺贝利核事故只有1个反应堆发生事故,在不考虑惰性气体前提下,福岛核事故泄漏放射性物质总量为520 PBq,大约为切尔诺贝利核事故总量的10%~15%(Steinhauser等, 2014),如果考虑放射性惰性气体(^{85}Kr 和 ^{133}Xe),福岛核事故泄漏放射性物质总量大于切尔诺贝利核事故。

对于泄漏放射性物质成分,Nishihara等(2012)对福岛核电站反应堆、乏燃料中的放射性核素清单进行计算,详细给出反应堆内16种裂变核素、10种锕系核素、2种活化元素活度总量(Nishihara等, 2012);Yamamoto在福岛核电站周边土壤测得Sr, Nb, Mo, Tc, Ru, Ag, Te, I, Ba, La, Pu, Am和Cm的同位素,并根据 $^{134}\text{Cs}/^{137}\text{Cs}$ 、 $^{238}\text{Pu}/^{239+240}\text{Pu}$ 比值推测反应堆的燃耗

情况,估算其他核素与 ^{137}Cs 的比值(Yamamoto, 2012);福岛核电站周边的树木和碎石中也检测到 ^3H , ^{14}C , ^{60}Co , ^{79}Se , ^{90}Sr , ^{99}Tc 和 ^{137}Cs 等核素(Tanaka等, 2014);Kirchner等(2012)认为事故时刻堆芯温度小于2700 K(Kirchner等, 2012),福岛核事故泄漏放射性核素主要是一些沸点较低的易挥发性核素,比如:惰性气体(^{85}Kr 和 ^{133}Xe)(Nitta等, 2014), ^{95}Zr - ^{95}Nb (Tagami等, 2011; Kanai, 2012), ^{99}Mo - ^{99m}Tc (Kanai, 2012), ^{110m}Ag (Kanai, 2012; Saegusa等, 2013; Lepage等, 2014; Saito等, 2014), ^{125}Sb (Saegusa等, 2013), ^{129}I (Miyake等, 2012), ^{131}I (Chino等, 2011), ^{133}I (Leon等, 2011), $^{134,135,136,137}\text{Cs}$ (Tagami等, 2011; Kanai, 2012; de Vismes Ott等, 2013; Zheng等, 2014), ^{129m}Te - ^{129}Te (Kanai, 2012; de Vismes Ott等, 2013), ^{132}Te - ^{132}I (Tagami等, 2011; Kanai, 2012; de Vismes Ott等, 2013; Tagami等, 2013), ^{140}Ba - ^{140}La (Tagami等, 2011; Kanai, 2012; de Vismes Ott等, 2013)等,而沸点较高的核素,比如 ^{59}Fe , ^{90}Sr , ^{95}Zr , ^{236}U 及 ^{239}Np , $^{239,240}\text{Pu}$, Am和Cm等超铀核素泄漏量较少(Tagami等, 2011; Shozugawa等, 2012; Zheng等, 2012; Schneider等, 2013; Steinhauser等, 2013; Zheng等, 2013; Sakaguchi等, 2014; Yamamoto等, 2014a),大部分留在反应堆内部(Schwantes等, 2012).但是部分研究表明事故期间日本气溶胶中仍然可以检测到福岛核事故产生的 $^{239+240}\text{Pu}$, ^{236}U (Shinonaga等, 2014),欧洲检测气溶胶中 $^{239+240}\text{Pu}$ 在福岛核事故期间轻微升

高,但是无法肯定是否来自福岛核事故(Lujanienė等, 2012). Tanabe(2012), Schwantes等(2012)和Le Petit等(2012)根据环境中核素的测量数据对堆芯情况与放射性核素泄漏进行分析(Blandford和Ahn, 2012; Le Petit等, 2012; Schwantes等, 2012; Tanabe, 2012).

因此,福岛核事故主要释放易挥发的放射性核素(放射性惰性气体, I, Cs和Te),而高沸点的核素(Sr, 锆系核素, Zr和Ce等)只有在核电站周边地区发现(Doi等, 2013),日本以外的北美、欧亚大陆等并未发现高沸点的核素,而切尔诺贝利核事故除了泄漏挥发性核素外,还包含大量的Sr, Ru, Zr, Ce, La, Pm和Pu等高沸点核素(Kashparov等, 2003).

福岛核事故排放的放射性物质80%进入太平洋,19%沉降于日本,不到1%沉降于北美、欧亚大陆等地区(Morino等, 2011; Stohl等, 2012b; Ten Hoeve和Jacobson, 2012; Yoshida和Kanda, 2012; Christoudias和Lelieveld, 2013),主要污染北太平洋;切尔诺贝利核事故只有不到10%量沉降于海洋且主要污染波罗的海、黑海等北欧海域,大部分放射性物质沉降于欧洲大陆(Evangelidou等, 2014). 由于日本独特的物理环境,大部分放射性物质进入海洋,福岛核事故对陆地污染范围较切尔诺贝利核事故的污染范围小(Steinhauser等, 2014),但是海洋环境的影响范围较大.

6 结论与展望

不考虑惰性气体排放,福岛核事故排放放射性物质总量大约为520 PBq,是切尔诺贝利核事故的10%左右,不足全球落下灰的千分之一;福岛核事故泄漏的放射性物质分布于日本、太平洋和世界其他地区的比重大约为19%, 80%和1%. 因此切尔诺贝利核事故仍然是至今为止最为严重的核事故,而福岛核事故是至今为止最为严重的海洋放射性污染核事故.

福岛核事故核素排放组成以低沸点的Kr, Xe, I, Cs和Te为主,高沸点Sr和Pu等核素大部分停留于反应堆内部,福岛核事故源项具有独特的放射性核素谱,与全球落下灰、切尔诺贝利核事故的核素谱存在

差异,核素活度或者原子数比值今后可作为一个典型的历史事件而记录于环境介质中. 日本附近的湖泊和海洋沉积物、陆地土壤中,来自福岛核事故的 ^{137}Cs 很可能可以作为今后独特的历史事件而被识别,并应用于地质年代学研究中,但是世界其他地区的沉积物中该事件将很难被识别.

目前对核事故中燃料棒、控制棒、压力壳和冷却剂等物理、化学等过程的了解还不充分,特别是淡水、海水与核燃料直接接触,放射性物质液态排放的总量与组成,核电站周边的非点源长期排放等仍然存在很大不确定性(Burns等, 2012). 事故后人为引入大量的冷却水与熔化后的核燃料仍然停留于核电站内部,并可能通过地下水进入环境中. 河流输入也可能作为一个长期的排放途径,把沉降于日本陆地的放射性物质搬运至近海海域. 核电站附近储存高放射性废水的废液罐也是潜在的污染源. 目前核电站附近海域放射性活度仍然居高不下,说明仍然存在持续的泄漏源项(Kanda, 2013). 寻找泄漏途径、定量泄漏通量、有效控制泄漏总量是日本今后一项长期而又艰巨的任务.

核事故后采取的冷却措施导致核电站周边储存着大量的高放射性的冷却水,日本联合美国、俄罗斯等国家正积极开展高放射性废液后处理的研究(Sylvester等, 2013),许多核素可以通过化学方法去除,但是 ^3H 无法通过化学方法去除,最终也很可能直接排入海洋,其他利益相关方或者国际组织应该及时建立环太平洋 ^3H 本底,以此关注日本方面采取的冷却水处理措施.

核事故的后果评价与环境修复是一项长期而复杂的任务,而源项的准确估计是最为基础的工作. 尽管国际上已经开展大量的研究工作,但是不同排放时间及空间的界定、不同来源的气象与水文数据的选取、不同模型的应用、不同来源的监测数据的校准等因素,都将导致福岛核事故的源项估计仍然存在很大不确定性. 今后仍然需要利用更多手段,从不同角度出发定量源项,减少源项评估过程中的不确定度,增加源项评估结果的可信度.

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